Fluid, Electrolyte, and Acid-Base Balance

Focus 7
Nursing 53A
Judy Ontiveros, RN, PHN, MSN



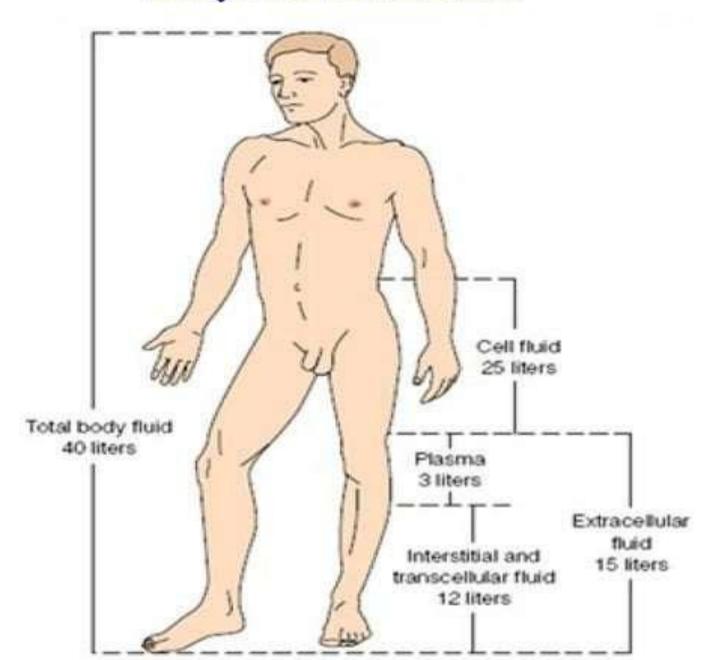
Objectives

- Describe how fluid volume and electrolytes are regulated to maintain homeostasis.
- Describe fluid and electrolyte balance.
 - Examine fluid volume imbalance and preventative interventions.
 - Define and discuss the major electrolytes in the maintenance of homeostasis
 - Explain the proper distribution of fluid in each body fluid compartment and the effect of mal-distribution
- Define and discuss the major electrolytes essential in the maintenance of homeostasis

Objectives

- Define populations particularly vulnerable to fluid and electrolyte imbalance.
- Discuss preventative measures to electrolyte imbalance.
- List common diagnostic tests related to fluid and electrolyte status.
- Discuss the nursing purpose responsibilities of each procedure.
- Examine altered means of fluid intake and the related nursing responsibilities/technical skills.

Body Fluid Distribution



Fluid and Electrolyte Balance

- Homeostasis
 - Balance of fluids, electrolytes, acids and bases
 - Physiologic processes that control intake and output
 - Body composed of 46 60% of adult weight
 - Every illness has potential to upset the balance

Maintaining Fluid Balance

- Intake should equal fluid output
 - Intake at 2000 mL
 - Output at 1500 mL
 - Difference within 200 300 mL

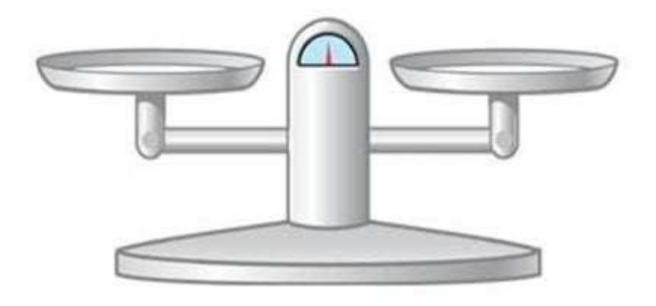
Fluid intake

Fluid output

2600-3000 mL

Ingested water	1200-1500 mL	Kidneys	1500 mL
Ingested food Metabolic oxidation	800-1100 mL 300 mL	Insensible loss through skin Insensible loss	600-800 mL
TOTAL	2600-3000 mL	through lungs Gastrointestinal tract	400-600 mL 100 mL
			CONTRACTOR OF

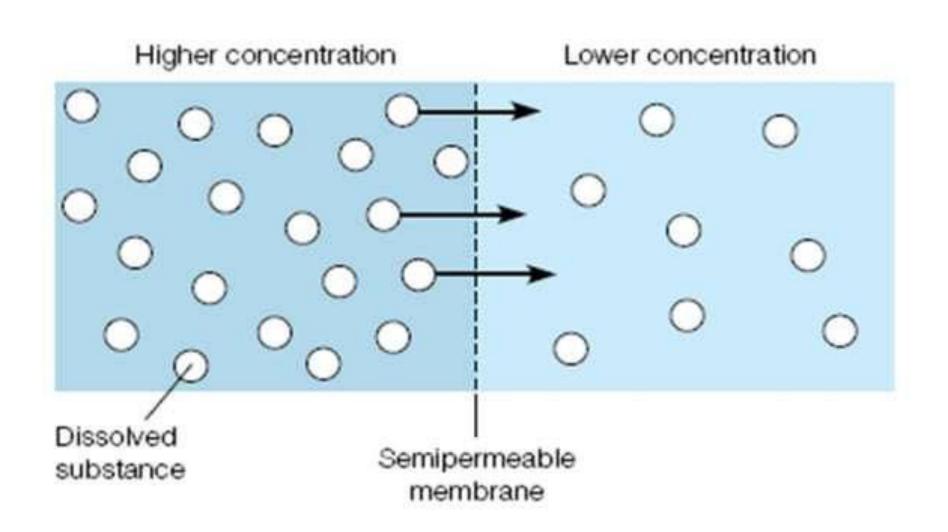
TOTAL



Movement of Fluids

- Osmosis
- Diffusion
- Filtration
- Active transport

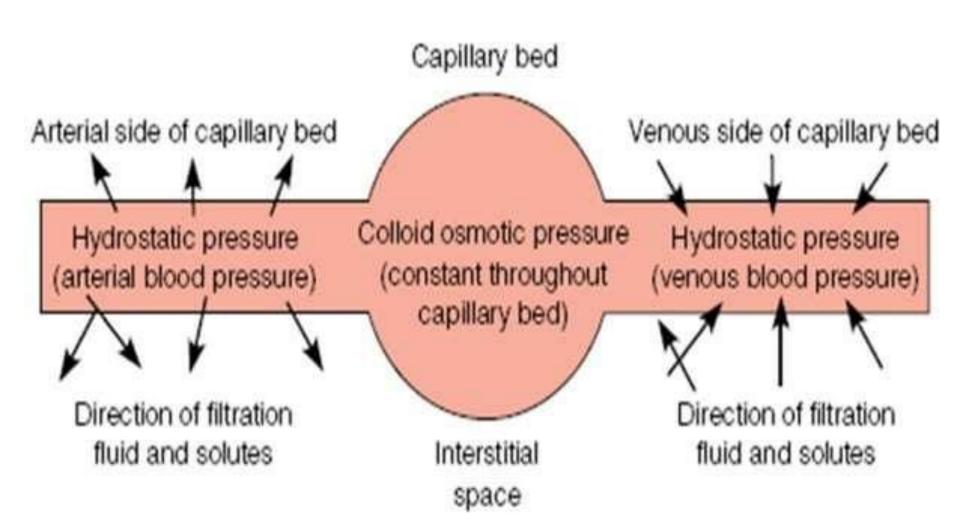
Diffusion



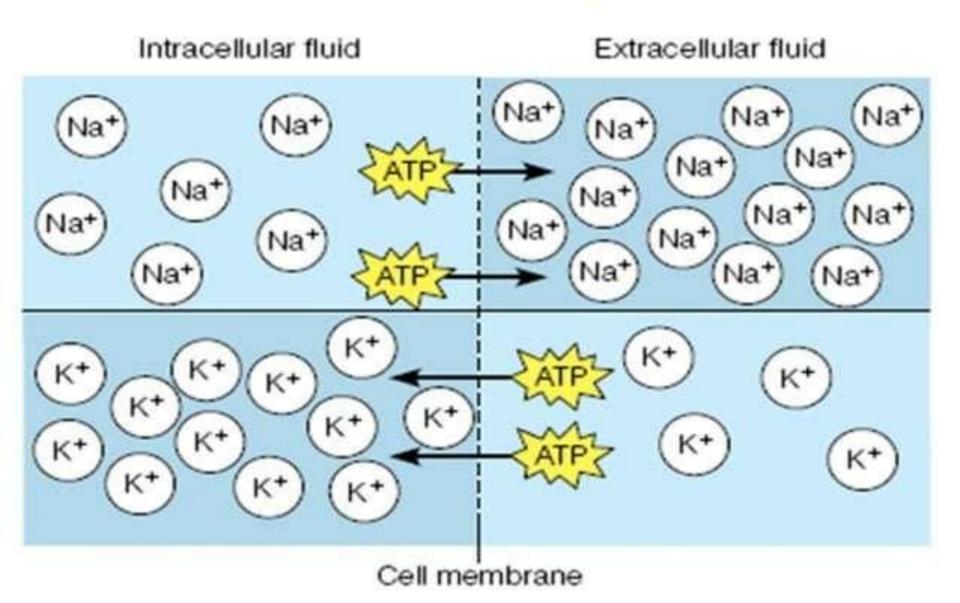
Osmosis

Higher concentration Lower concentration Water Dissolved Semipermeable substances membrane molecules

Filtration



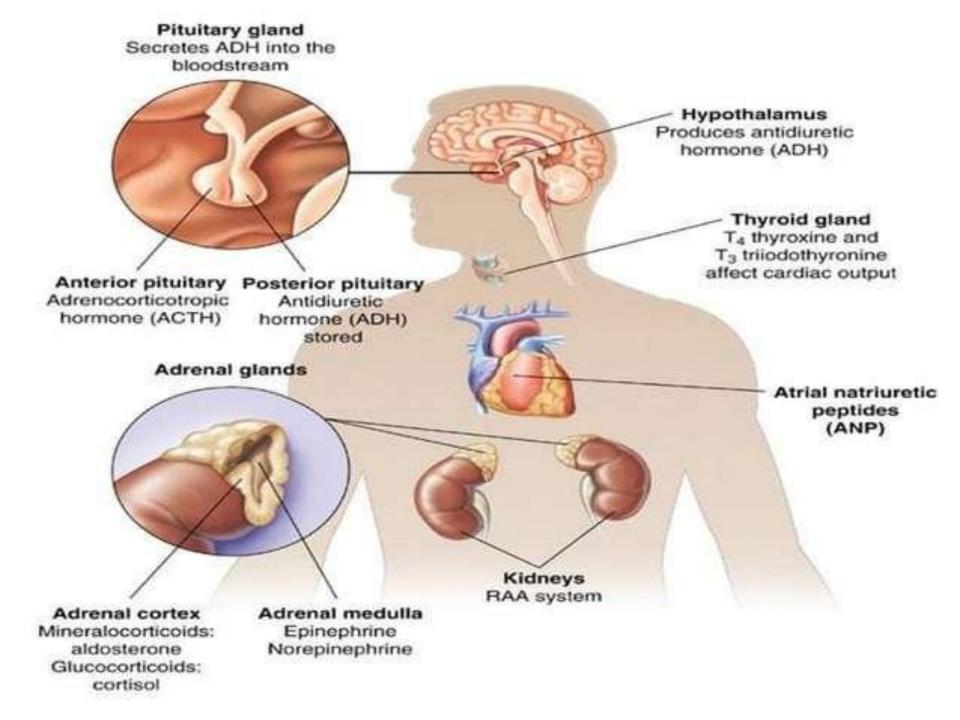
Active Transport

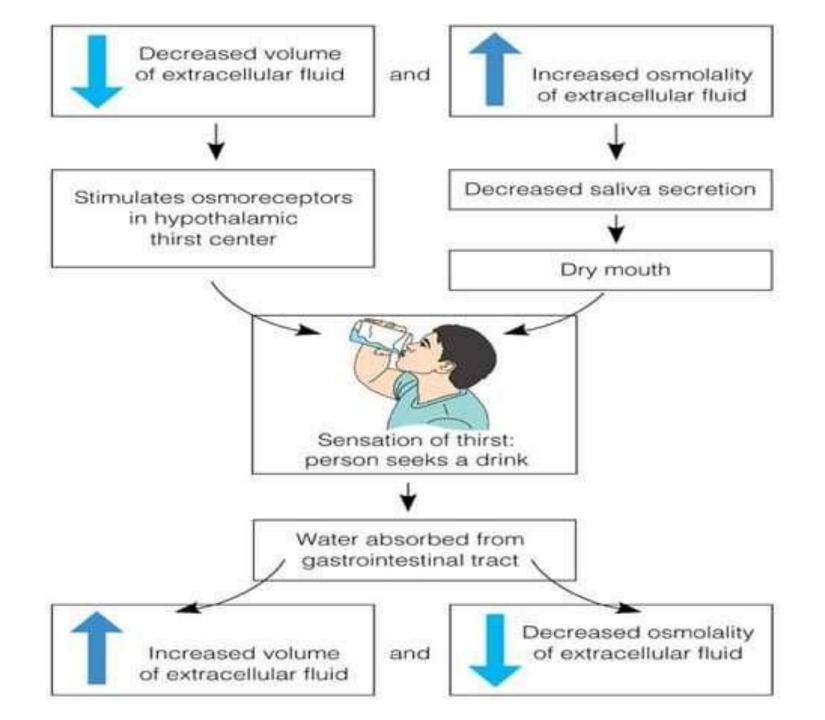


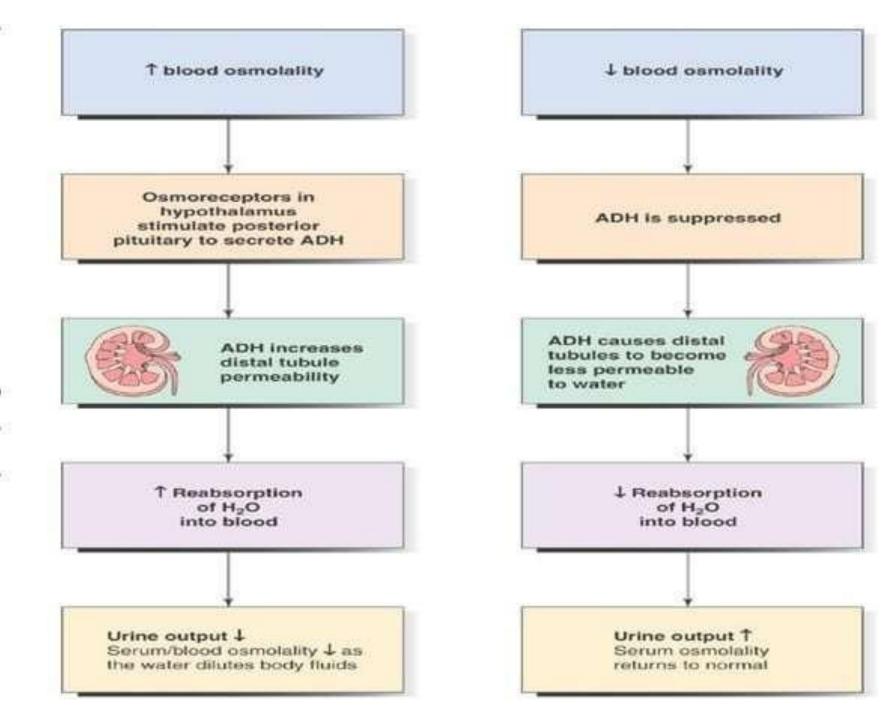
Regulating Body Fluids

- Fluid intake
 - Thirst
- Fluid output
 - Urine
 - Insensible loss
 - Feces

- Maintaining homeostasis
 - Kidneys
 - ADH
 - Renin-angiotensinaldosterone system
 - Atrial natriuretic system







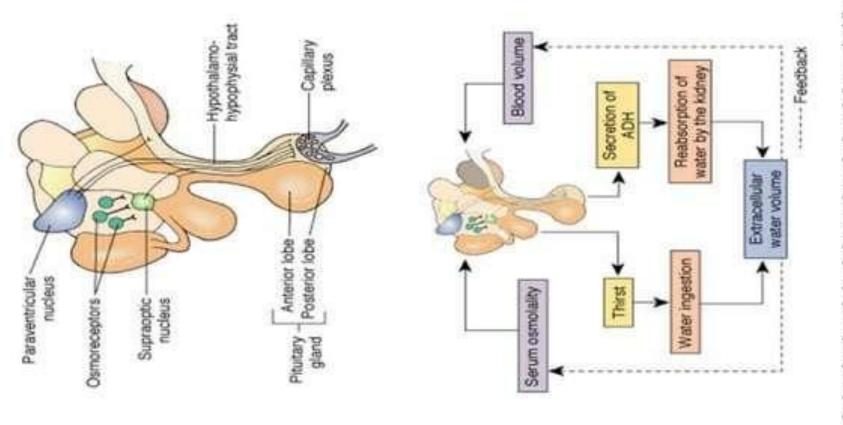


Figure 33-10 (Top) Sagittal section through the pituitary and anterior hypothalamus. Antidiuretic hormone (ADH) is pituitary, where it can be released into the blood. (Bottom) Pathways for regulation of extracellular water volume by formed primarily in the supraoptic nucleus and to a lesser extent in the paraventricular nucleus of the hypothalanus. It is then transported down the hypothalamohypophysial tract and stored in secretory granules in the posterior thirst and antidiuretic hormone.

Fluid and Electrolyte Balance

- Volume imbalance:
 - Net volume gain
 - hypervolemia

- Net volume loss
 - hypovolemia

- Water intoxication
 - overhydration

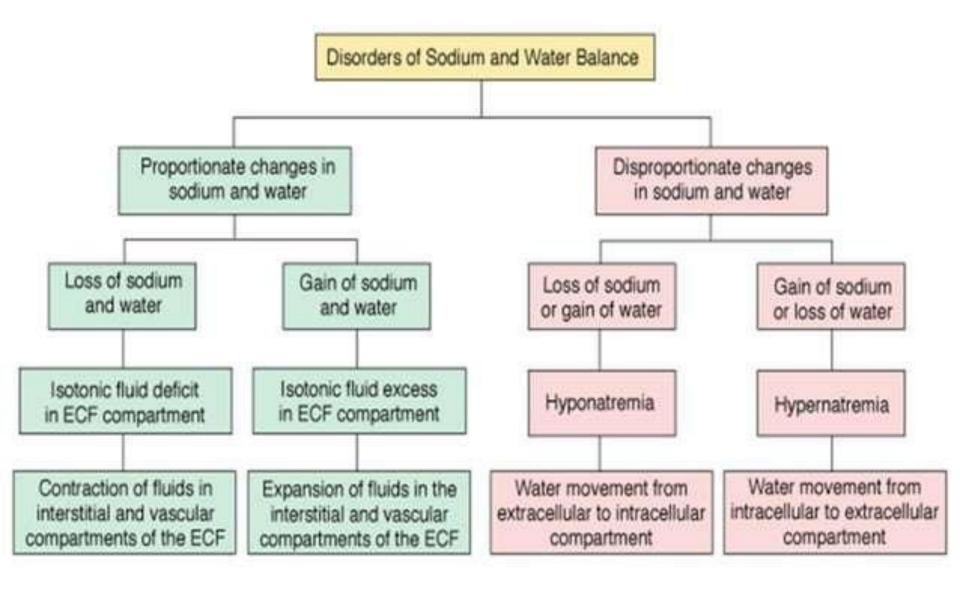


Figure 33-7 The effect of proportionate and disproportionate changes in sodium and water balance on extracellular sodium concentration.

Fluid Volume Excess (FVE) Intake Exceeds Output

- Weight gain
 - Surgery
- Pitting edema
- Cough
- Dyspnea

- Cardiac palpitations
- Decreased urinary output
- Mental status changes

Fluid Volume Deficit (FVD) Output exceeds intake

- Diarrhea
- Diuretics
- Vomiting
- Gastric suction
- Anorexia
- Increasing fatigue and weakness
- Weight loss

- Fever
- Excess urine output
- Change in mental status
- Traumatic injury (burn)
- Blood loss

Factors Affecting Distribution

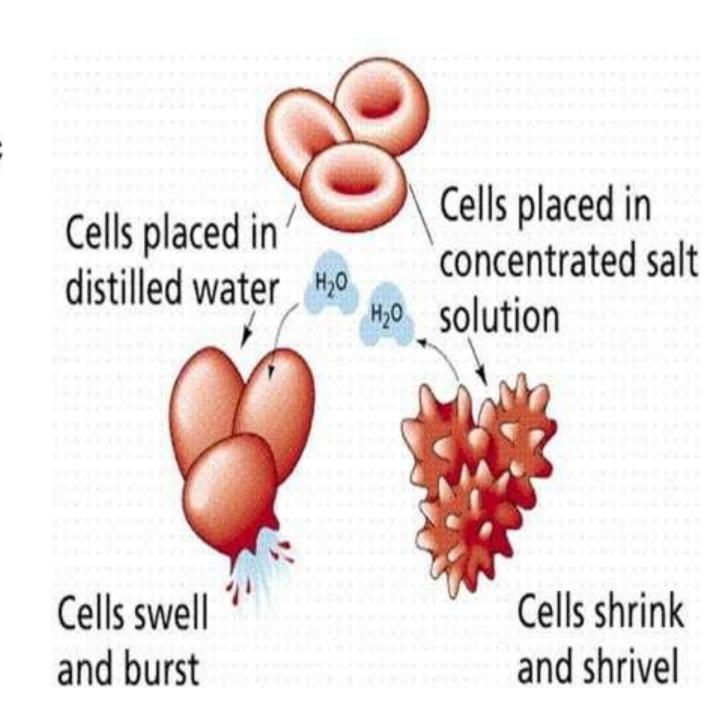
Arterial Venous Filtration end end Capillary Capillary filtration colloidal osmotic pressure pressure Hydrostatic Interstitial Tissue Pressure hydrostatic fluid pressure pressure Osmotic Pressure Lymph channels

Figure 33-5 Exchange of fluid at the capillary level.

Hypertonic

Isotonic

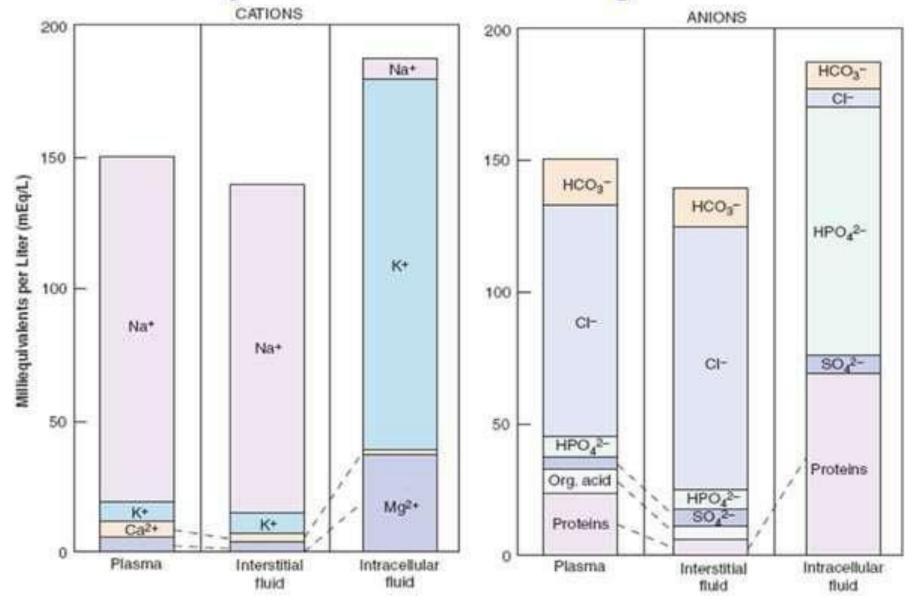
Hypotonic



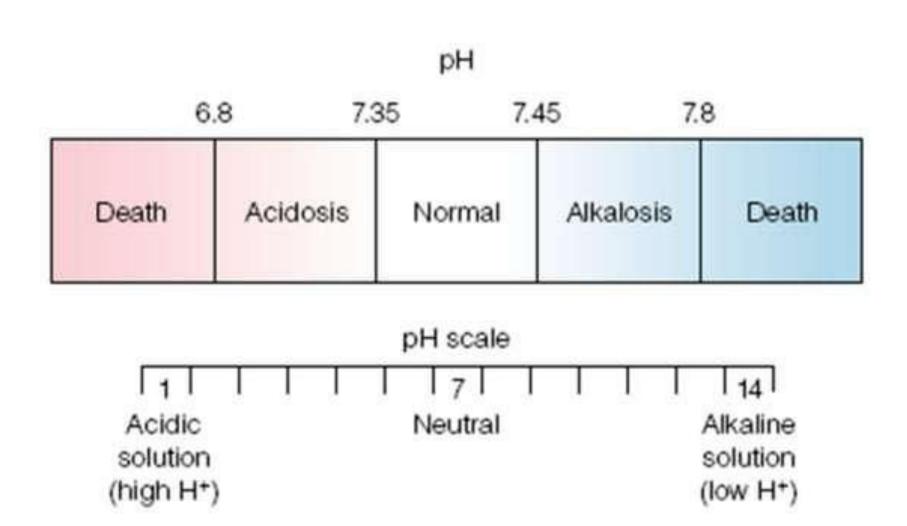
Regulating Electrolytes

- Sodium
- Potassium
- Calcium
- Magnesium
- Chloride
- Phosphate
- Bicarbonate

Composition of Body Fluids



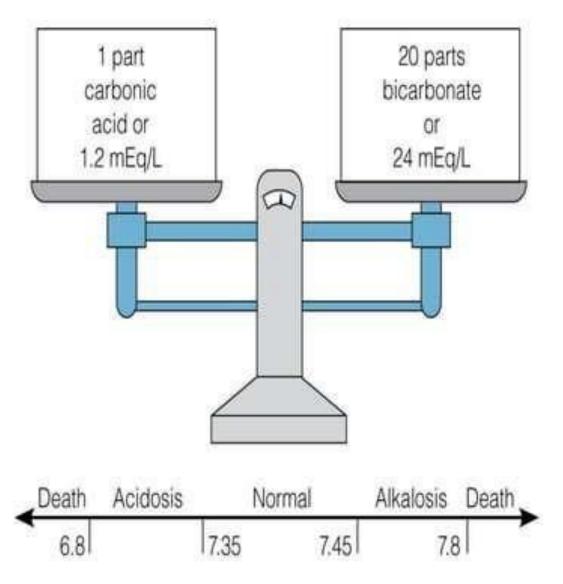
Regulating Acid-Base Balance



Regulating Acid-Base Balance

- Low pH = acidic
- High pH = alkalinic
- Body fluids maintained between pH of 7.35 and 7.45 by
- Buffers
- Respiratory system
- Renal system

Figure 52-10 Carbonic acid-bicarbonate ratio and pH.

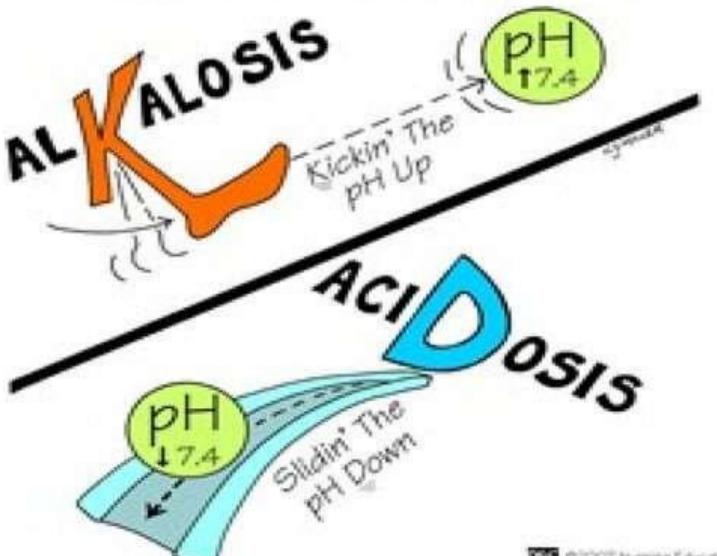


- Prevent excessive changes in pH
- Major buffer in ECF is HCO3 and H2CO3
- Other buffers include:
 - Plasma proteins
 - Hemoglobin
 - Phosphates

Acid-Base Imbalances

- Respiratory acidosis
- Respiratory alkalosis
- Metabolic acidosis
- Metabolic alkalosis

ACIDOSIS - ALKALOSIS



ACID BASE MNEMONIC (ROME)



Respiratory



Opposite

pH↑ PCO2 ↓ Alkalosis

pH ↓ PCO2 ↑ Acidosis



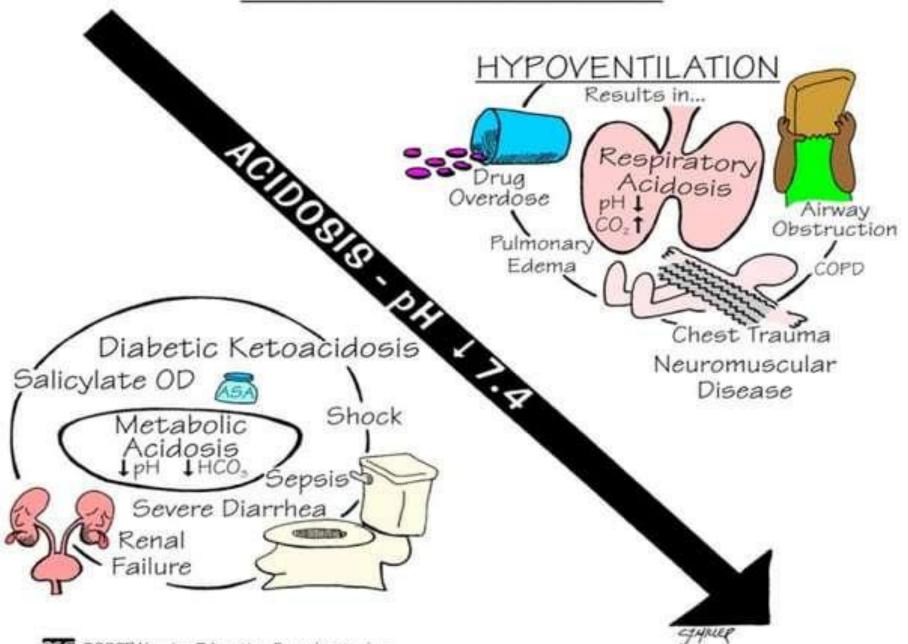
Metabolic



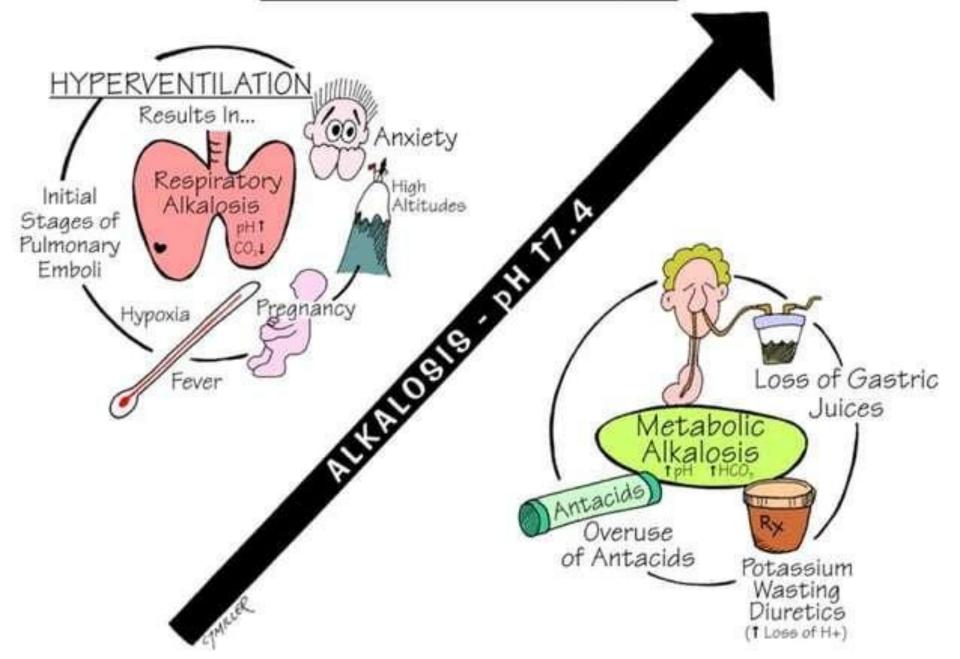
Equal

pH \uparrow HCO₃ \uparrow Alkalosis pH \downarrow HCO₃ \downarrow Acidosis

CAUSES OF ACIDOSIS



CAUSES OF ALKALOSIS



Lungs

- Regulate acid-base balance by eliminating or retaining carbon dioxide
- Does this by altering rate/depth of respirations
- Faster rate/more depth = get rid of more CO2 and pH rises
- Slower rate/less depth = retain CO2 and pH lowers

RESPIRATORY ACIDOSIS

Hypoventilation → Hypoxia

 Rapid, Shallow Respirations

> ↓BP with Vasodilation

• Dyspnea //

Headache

Hyperkalemia

 Dysrhythmias (↑K) I can't catch my breath.

> Drowsiness, Dizziness, Disorientation

 Muscle Weakness, Hyperreflexia

Causes:

↓Respiratory Stimuli (Anesthesia, Drug Overdose) COPD

Pneumonia Atelectasis

Retention of CO₂ by Lungs

TpCO₂

(†45mm Hg)

, pH

(17.35)

RESPIRATORY ALKALOSIS

LpCO2

(35mm Hg)

Loss of

CO2 from Lungs

₽pH

 $(\uparrow 7.45)$

Seizures

Deep, Rapid
 Breathing

Hyperventilation

Tachycardia

↓or Normal BP

Hypokalemia

Numbness
 & Tingling of Extremities

Lethargy & Confusion

Light Headedness

Nausea, Vomiting

Causes:

Hyperventilation
(Anxiety PF Fea

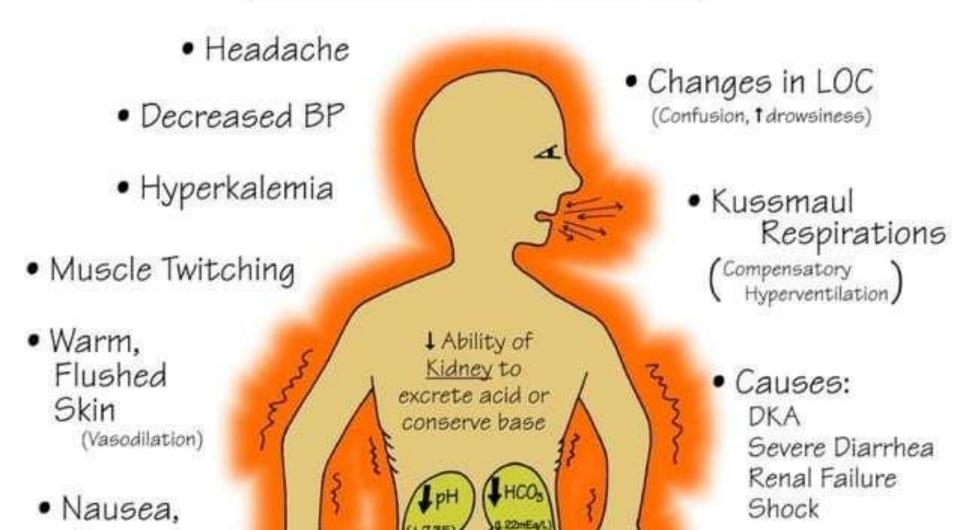
(Anxiety, PE, Fear)

Mechanical Ventilation

Kidneys

- Regulate by selectively excreting or conserving bicarbonate and hydrogen ions
- Slower to respond to change

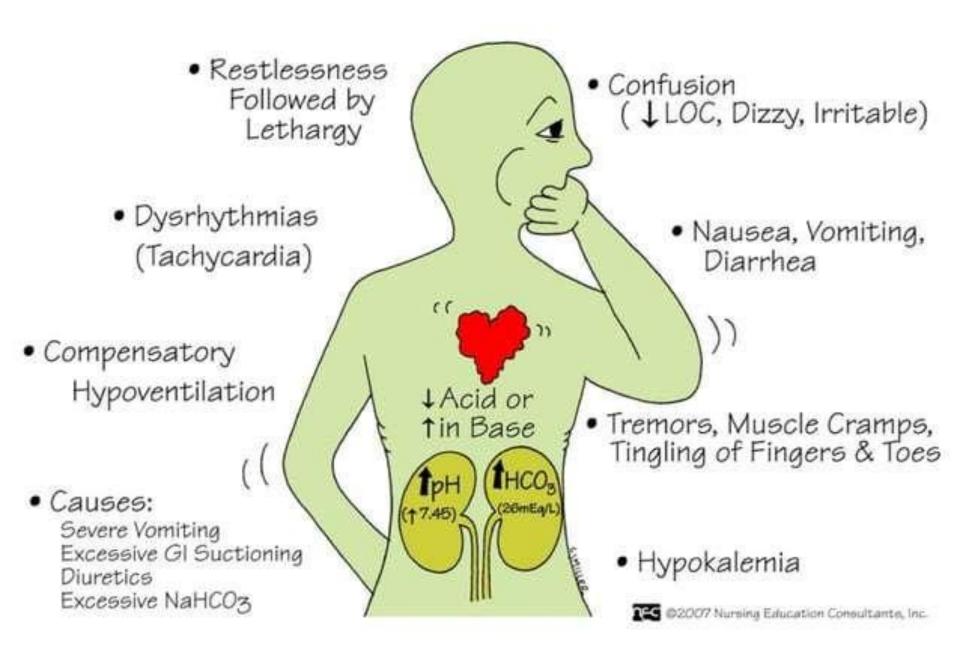
METABOLIC ACIDOSIS



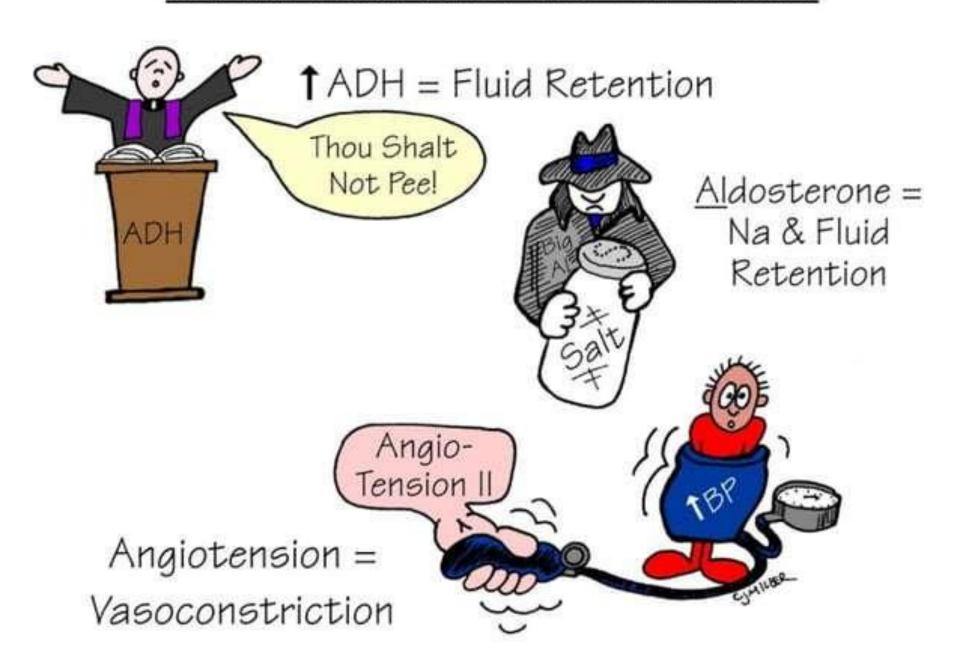
Vomiting

Diarrhea

METABOLIC ALKALOSIS



RENAL COMPENSATION IN SHOCK



Arterial Blood Gas

PH 7.35-7.45

PaO2 80-100 mm Hg

PaCO2 35-45 mm Hg

HCO3 22-26 mEq/L

Base Excess -2 - +2 mEq/L

SaO2 95-98%

Factors Affecting Body Fluid, Electrolyte, and Acid-Base Balance

- Age
- Gender
- Body size
- Environmental temperature
- Lifestyle

Risk Factors for Fluid, Electrolyte, and Acid-Base Imbalances

- Chronic diseases
- Acute conditions
- Medications
- Treatments
- Extremes of age
- Inability to access food and fluids

Specific Illnesses

- COPD, Asthma,
 Cystic Fibrosis
- CHF
- Kidney disease
- Diabetes Mellitus
- Cushing's or Addison's disease
- Cancer

- Malnutrition, anorexia nervosa, bulimia
- Ileostomy
- Gastroenteritis
- Bowel obstruction
- Head injury
- Fever, draining wounds, fistulas
- Surgery

Medications to Monitor

- Diuretics
 - Water depletion
 - Electrolyte depletion
- Corticosteroids
 - Water retention
- NSAIDS/Opoiods
 - Constipation

Treatments that affect fluid balance

- Chemotherapy
- IV therapy or TPN
- Nasogastric suction
- Enteral Feedings
- Mechanical Ventilation

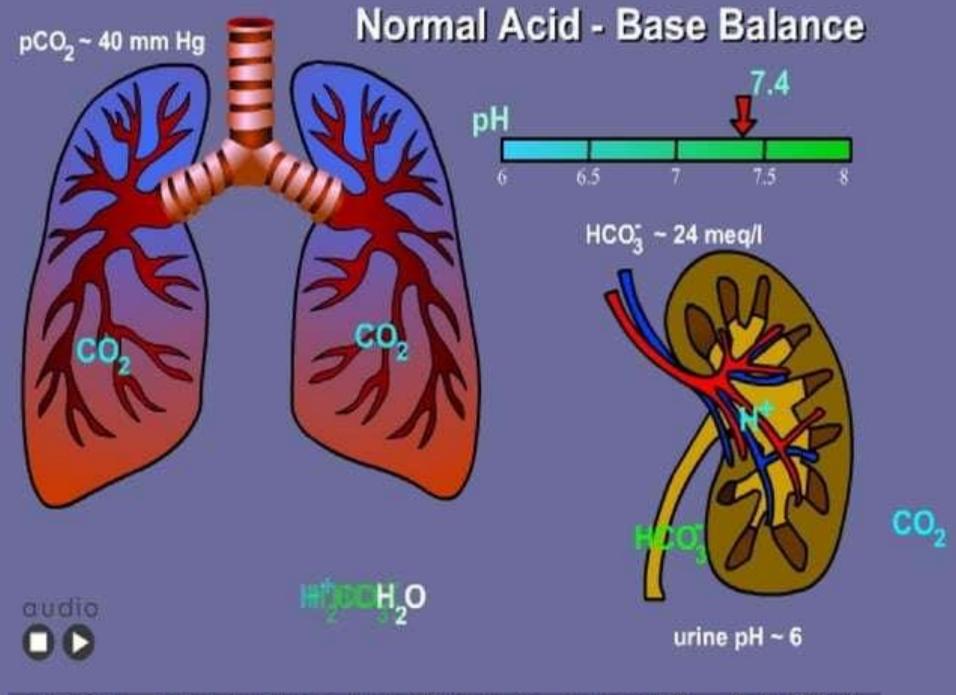
Fluid Imbalances

- Isotonic loss of water and electrolytes (fluid volume deficit)
- Isotonic gain of water and electrolytes (fluid volume excess)
- Hyperosmolar loss of only water (dehydration)
- Hypo-osmolar gain of only water (overhydration)

Electrolyte Imbalances

- Hyponatremia
- Hypernatremia
- Hypokalemia
- Hyperkalemia
- Hypocalcemia
- Hypercalcemia

- Hypomagnesemia
- Hypermagnesemia
- Hypochloremia
- Hyperchloremia
- Hypophosphatemia
- Hyperphosphatemi a



equilibrium metabolic acidosis metabolic alkalosis resp. acidosis resp. alkalosis

Hypernatremia

Sodium

- 145 mEq/L

Risk Factors

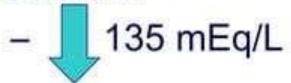
 water deprivation, hypertonic tube feedings, with inadequate water supplements, fever, unusually high Na+ intake, profuse sweating, heatstroke, diabetes insipidus

Defining Characteristics

 thirst, dry sticky mucus membranes, red, dry swollen tongue, dyspnea, fatigue, disorientation, convulsions

Hyponatremia

Sodium



Risk Factors

 loss of GI fluids, adrenal insufficiency, sweating, diuretics, gain of water, hypotonic tube feedings, oral ingestion of water, excess administration of Dextrose, disease states head injury, AIDS, malignant tumors

Defining Characteristics

headache, muscle twitching, seizure activity, abdominal cramps,
 N&V, anorexia, lethargy, confusion

Hyperkalemia

- Potassium
- 5.5 mEq/L (serum)

Risk Factors

 decrease K+ excretion, renal failure, K+ conserving diuretics, hypoaldosteronism, High K+ intake during renal insufficiency, too rapid K+ infusion, too high dose of K+ PCN or KCL dose

Defining Characteristics

ECG changes, vague muscle weakness, GI hyperactivity

 diarrhea, nausea, irritability, apathy, confusion, cardiac
 arrhythmias, arrest, absence of reflexes, flaccid muscles
 paralysis, numbness and paresthesias in extremities, face and
 tongue

Hypokalemia

- Potassium
- 3.5 mEq/L (serum)
- Risk Factors
 - vomiting, gastric suction, diarrhea, heavy diaphoresis, use of K+ wasting drugs like lasix and thiazide diuretics, steriod adminstration, low intake of K+ from ETOHism or anorexia, hyperaldosteronism, osmotic diuresis from uncontrolled DM
- Defining Characteristics
 - ABGs show alkalosis, ECG changes, cardiac arrhythmias, muscle weakness, paresthesias, muscle soreness, leg cramps, fatigue, lethargy, anorexia, N&V, decrease bowel sounds and motility,



- Muscle Twitches → Cramps → Paresthesia
- Irritability & Anxiety
- * 18P
- * EKG Changes
- * Dyorhythmias Irregular Rhythm
- Abdominal Cramping
- * Diarrhea





EKG Changes with K⁺ Imbalance

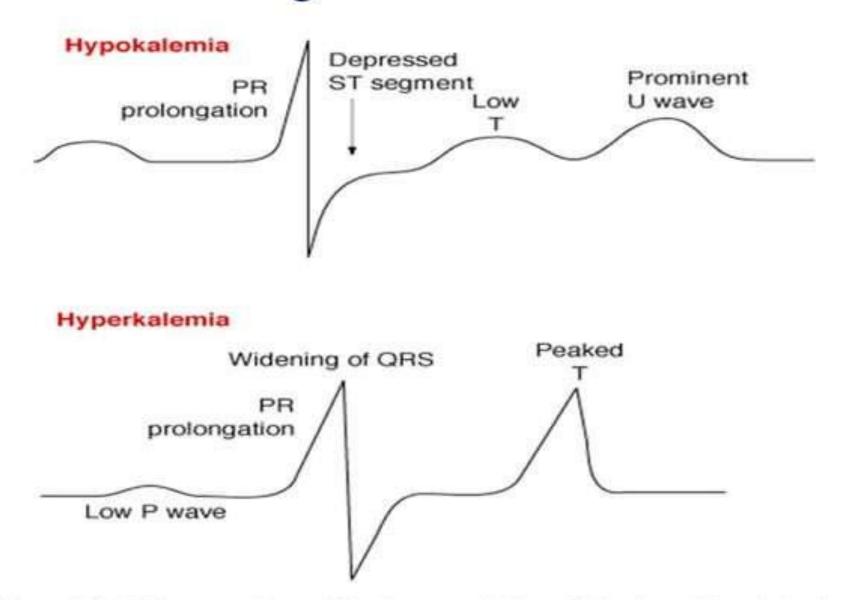


Figure 33-13 Electrocardiographic changes with hyperkalemia and hypokalemia.

Hypokalemia





- * Alkalosis
- * Shallow Respirations
- * Irritability
- * Confusion, Drowsiness
- * Weakness, Fatigue
- * Arrhythmias Irregular rate, Tachycardia
- * Lethargy
- * Thready Pulse
- ★ Intestinal Motility Nausea Vomiting Ileus



POTASSIUM LEVELS IN ACIDOSIS



Potassium Goes Up In Acidosis

Hypercalcemia

Calcium

- 15.5 mg/dL

Risk Factors

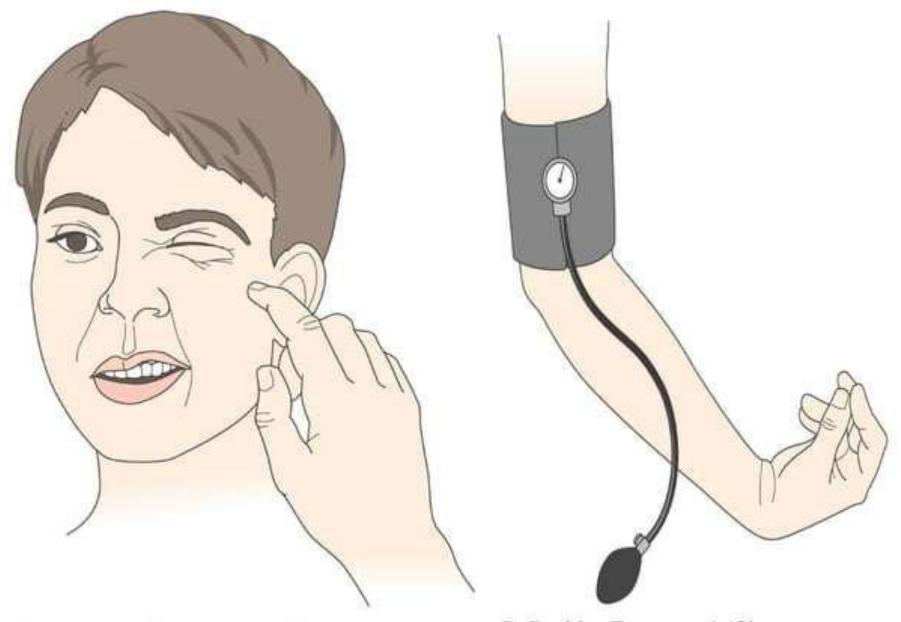
 hyperparathyroidism, prolonged immobility, megadoses of vitamin D, bone malignancy, paget's disease, thiazide diuretics, overuse of Ca+ containing antacids and supplements

Defining Characteristics

 ECG changes, lethargy, weakness, depressed deep tendon reflexes, constipation, anorexia, N&V, polyuria, polydipsia, decreased memory and attention span, confusion, renal stones, flank pain, neuroses, psychosis reversible, cardiac arrest

Hypocalcemia

- Calcium
- 4.5 mEq/L
- R k Factors
 - hypoparathyroidism, malabsorption, vitamin D deficiency, acute pancreatitis, thyroid CA, hypomagnesemia, alkalosis, sepsis, alcohol abuse
- Defining Characteristics
 - numbness, tingling, hyperactive deep tendon reflexes, muscle tremors, cramps, progressing to tetany and convulsions, cardiac arrhythmias, mental changes, Chvostek's sign, Trousseau's sign



A. Positive Chvostek's Sign

B. Positive Trousseau's Sign

Collecting Assessment Data

- Nursing history
- Physical assessment
- Clinical measurement
- Review of laboratory test results
- Evaluation of edema

Figure 52-11 Evaluation of edema.

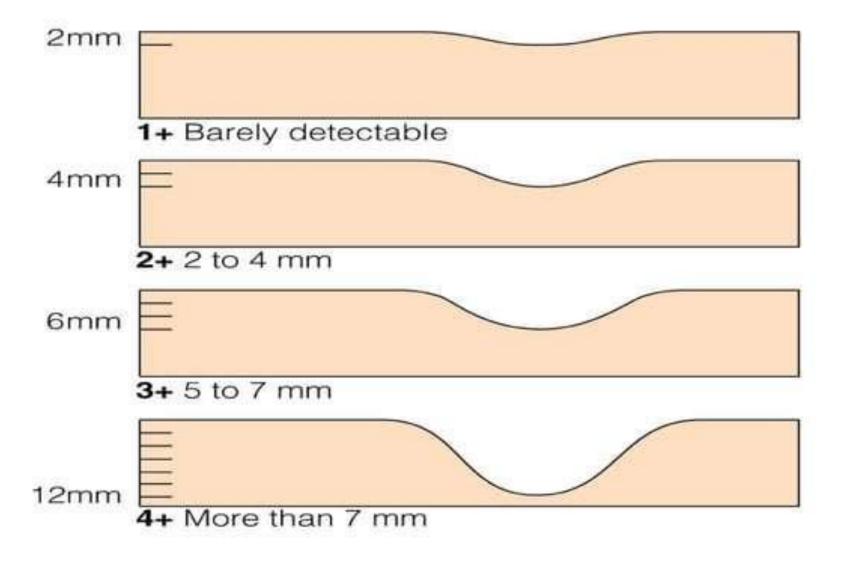
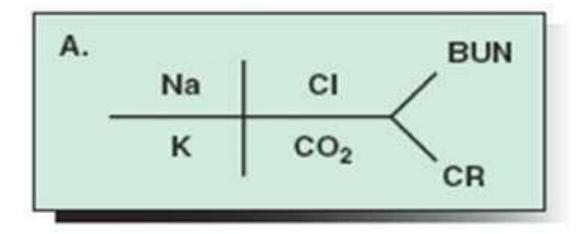
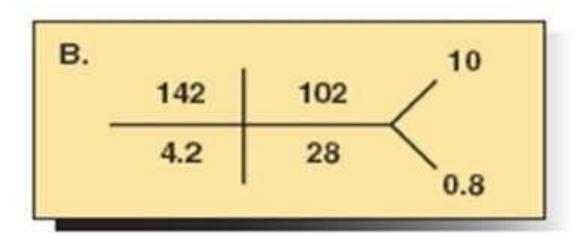


Diagram of Serum Electrolyte Results





Desired Outcomes

- Maintain or restore normal fluid balance
- Maintain or restore normal balance of electrolytes
- Maintain or restore pulmonary ventilation and oxygenation
- Prevent associated risks
- Tissue breakdown, decreased cardiac output, confusion, other neurologic signs

Nursing Interventions

- Monitoring
- Fluid intake and output
- Cardiovascular and respiratory status
- Results of laboratory tests
- Assessing
- Client's weight
- Location and extent of edema, if present
- Skin turgor and skin status
- Specific gravity of urine
- Level of consciousness, and mental status

Nursing Interventions

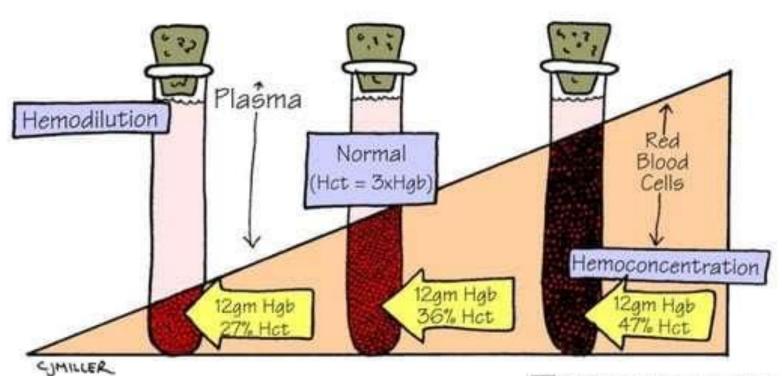
- Fluid intake modifications
- Dietary changes
- Parenteral fluid, electrolyte, and blood replacement
- Other appropriate measures such as:
- Administering prescribed medications and oxygen
- Providing skin care and oral hygiene
- Positioning the client appropriately
- Scheduling rest periods

- Monitoring daily Intake and Output (I&O)
- Monitoring daily Weight -same time (am), same clothes
 - 1 Kg.(2.2 lbs) is equal to 1 Liter(1000 mL) of fluid loss or gain
- Orthostatic Blood Pressure monitoring
 - L____ S____ St____

Laboratory tests-

- Electrolytes- Metabolic Panel
 - Serum Na
 - Serum K
 - Blood Urea Nitrogen (BUN)
 - Creatinine
 - Glucose
- Urine osmolality (Sp. Gravity)
- H & H or CBC

USING Hgb & Hct AS A GUIDE TO HYDRATION STATUS



Implementing and Documenting I & O

- Evaluating Outcomes
 - HCT (hematocrit)
 - Range of 40 54% males
 - Range of 38 47% Females
 - Increased values = FVD
 - Decreased Values = FVE
 - Specific Gravity 1.010-1.-25
 - High = FVD
 - Low = FVE

Promoting Fluid and Electrolyte Balance

- Consume 6-8 glasses water daily
- Avoid foods with excess salt, sugar, caffeine
- Eat well-balanced diet
- Limit alcohol intake
- Increase fluid intake before, during, after strenuous exercise
- Replace lost electrolytes

Promoting Fluid and Electrolyte Balance

- Maintain normal body weight
- Learn about, monitor, manage side effects of medications
- Recognize risk factors
- Seek professional health care for notable signs of fluid imbalances

Practice Guidelines Facilitating Fluid Intake

- Explain reason for required intake and amount needed
- Establish 24 hour plan for ingesting fluids
- Set short term goals
- Identify fluids client likes and use those
- Help clients select foods that become liquid at room temperature
- Supply cups, glasses, straws
- Serve fluids at proper temperature
- Encourage participation in recording intake
- Be alert to cultural implications

Planning and Implementation of I & O

- Assessing each patient's situation
 - Age
 - Infants and Children
 - Greater fluid turnover high metabolic rate
 - Kidneys immature lose more water
 - Respirations rapid –
 - Body surface area larger than adult Increases insensible losses
 - Fluid and electrolyte losses occur very rapidly
 - Elderly
 - Thirst response diminished
 - Nephrons less functional to conserve water
 - Increase risk of dehydration
 - Risks of HD, CRF, multiple medications
 - increases risk for fluid and electrolyte imbalance

Practice Guidelines Restricting Fluid Intake

- Explain reason and amount of restriction
- Help client establish ingestion schedule
- Identify preferences and obtain
- Set short term goals; place fluids in small containers
- Offer ice chips and mouth care
- Teach avoidance of ingesting chewy, salty, sweet foods or fluids
- Encourage participation in recording intake

Correcting Imbalances

- Oral replacement
 - If client is not vomiting
 - If client has not experienced excessive fluid loss
 - Has intact GI tract and gag and swallow reflexes

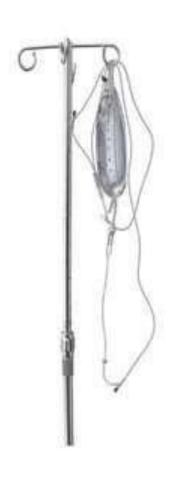
Correcting Imbalances

- Restricted fluids may be necessary for fluid retention
- Vary from nothing by mouth to precise amount ordered
- Dietary changes

Oral Supplements

- Potassium
- Calcium
- Multivitamins
- Sports drink

Correcting Imbalances



Intravenous Fluids

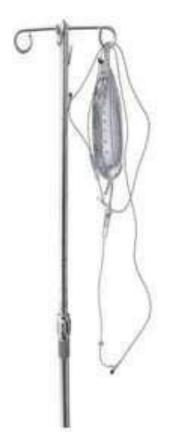
Hypotonic

- osmotic pressure less to that of plasma.
- 0.45% Na CL or half normal saline
- used for dehydration and
- promotes waste elimination of kidneys

Intravenous Fluids



- osmotic pressure equal to that of plasma
- Example #1 5% dextrose in H2O (D5W)
 - supplies free water to aid in renal excretion of solutes
 - expands intracellular and extracellular volumes
- Example #2 Lactated Ringers,
 0.9% NaCL or normal saline (NS)
 - expands vascular volume
 - contains multiple electrolytes in physiological concentrations
 - used to treat hypovolemia, burns, and diarrhea
 - used to treat mild metabolic acidosis



Intravenous Fluids

Hypertonic

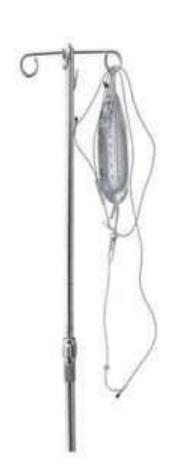
osmotic pressure above that of plasma



Example #1 - 5% dextrose in 0.45% NaCl (D5 1/2 NS)

- treats hypovolemia
- maintains hydration
- draws fluid out of the intracellular and interstitial spaces into the vascular space
- expands volume

Intravenous Fluids



- Hypertonic
 - Example #2 5% dextrose 0.9%NaCl (D5NS)
 - replaces calories and electrolytes
 - temporary treatment of hypovolemia

Evaluation

- Collect data as identified in the plan of care
- If desired outcomes are not achieved, explore the reasons before modifying the care plan