

# ANALYTICAL TOXICOLOGY



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Msc-2

Roll no-16

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## ***Urine***

Urine is useful for screening tests as it is often available in large volumes and usually contains higher concentrations of drugs or other poisons than blood. The presence of metabolites may sometimes assist identification if chromatographic techniques are used. A 50-ml specimen from an adult, collected in a sealed, sterile container, is sufficient for most purposes; no preservative should be added. The sample should be obtained as soon as possible, ideally before any drug therapy is initiated.

## ***Stomach contents***

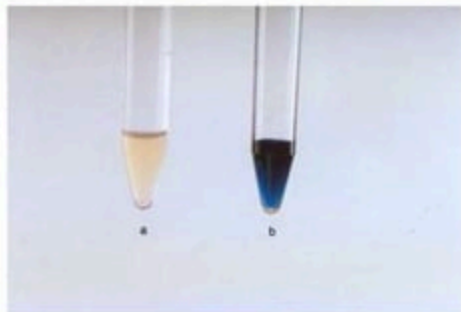
Stomach contents may include vomit, gastric aspirate and stomach washings — it is important to obtain the first sample of washings, since later samples may be very dilute. A volume of at least 20 ml is required to carry out a wide range of tests; no preservative should be added. This can be a very variable sample and additional procedures such as homogenization followed by filtration and/or centrifugation may be required to produce a fluid amenable to analysis.

## ***Blood***

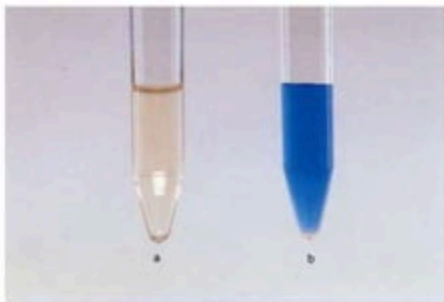
Blood (plasma or serum) is normally reserved for quantitative assays but for some poisons, such as carbon monoxide and cyanide, whole blood has to be used for qualitative tests. For adults, a 10-ml sample should be collected in a heparinized tube on admission.



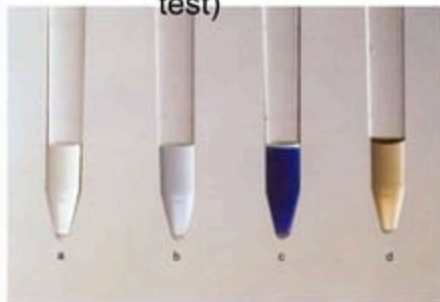
Phenothiazines  
(FPN test)



Imipramine  
(Forrest  
test)

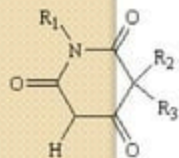
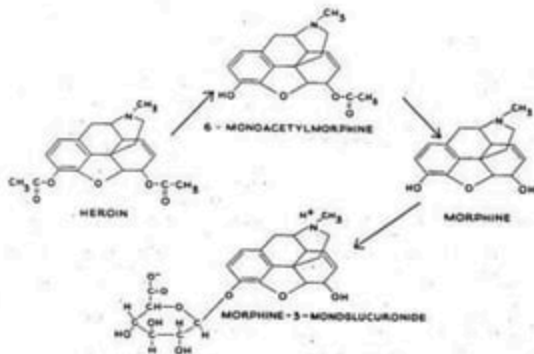
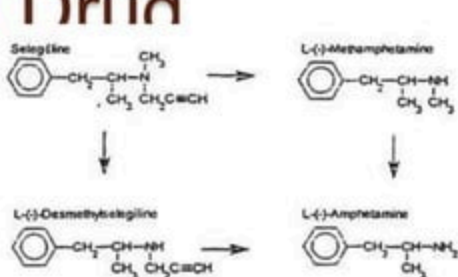


Paracetamol and  
phenacetin  
(ammonia test)

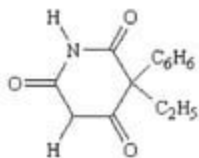


Paraquat and diquat  
(dithionite test)

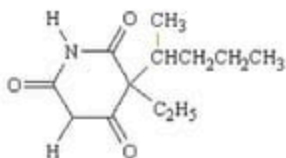
# Analysis of Metabolites of Drugs



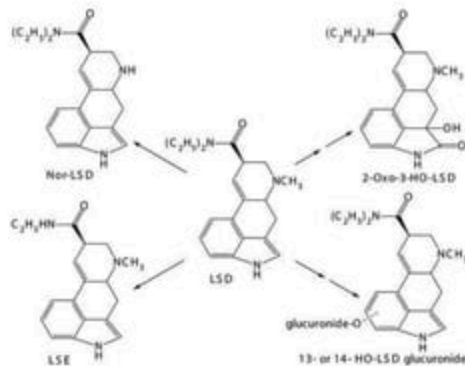
Barbituates

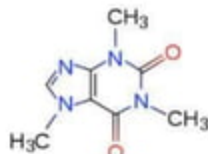


Phenobarbital

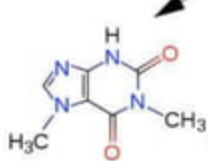


Pentobarbital

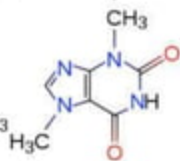




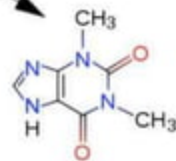
**Caffeine**



**Paraxanthine**  
(84%)



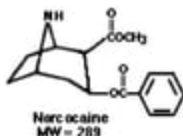
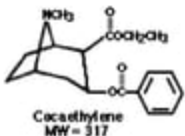
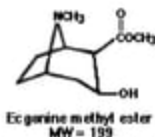
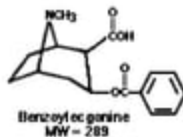
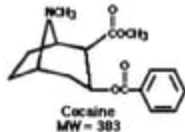
**Theobromine**  
(12%)



**Theophylline**  
(4%)

**Table 15-4**  
**Urinary Metabolites of Benzodiazepines**

Generic name	Brand name	Urinary metabolite
Oxazepam	Serax	Oxazepam
Temazepam	Restoril	Temazepam, Oxazepam
Chlordiazepoxide	Librium	Nordiazepam, Oxazepam
Diazepam	Valium	Diazepam, Nordiazepam, Oxazepam, Temazepam
Prazepam	Centrax, Verstran	Nordiazepam, Oxazepam
Clorazepate	Tranxene	Nordiazepam, Oxazepam
Medazepam	Nobrium	Nordiazepam, Oxazepam, Temazepam
Alprazolam	Xanax	$\alpha$ -Hydroxyalprazolam
Clonazepam	Klonopin	Aminoclonazepam, Clonazepam



# Sample Treatment

- **Protein Precipitation (PP)**
  - It is accomplished by using organic solvent (typically acetonitrile or methanol) or an acid (typically perchloric or trichloroacetic acid). It is followed by centrifugation to separate proteins from liquid supernatant.
- **Liquid-liquid extraction (LLE)**
  - To obtain a sensitive analysis for a complex biological media (plasma, urine) liquid-liquid extraction (LLE).
  - LLE is in general simpler also less expensive and flexible as several samples may be prepared in parallels.
- **Solid-phase extraction (SPE)**
  - Higher recoveries, no problems with emulsions, less solvent consumption and a smaller sample volume requirement.
  - Sample treatment with high speed and feasibility for treatment of numerous samples at one time is possible.

# Metabolite identification

*Full scan*

*Precursor ion and constant  
neutral loss scan*

**Product ion scan**

**MRM**



# Metabolite Determination

- LC-MS/MS
- *Reversed phase chromatography*
  - Reversed phase chromatography is most widely used technique in analysis of drugs and their metabolites due to its extensive application to most small molecules which are separated by their degree of hydrophobic interaction with the stationary phase.
  - An increased polarity of the metabolite decreased retention on the stationary phase.
  - For polar metabolites short chain bonded phases, such as C<sub>8</sub>, phenyl or cyano are more appropriate. Add ion-pairing reagent into mobile phase.
- *Ultra-high performance liquid chromatography (UHPLC)*
  - For fast analyses using sub-2 $\mu$ m particle column dimensions are typically 50x2 mm. An additional benefit of UHPLC is the low consumption of mobile phase, where it saves at least 80% compared to HPLC.
  - Advantages as enhanced separation efficiency, short analysis time and high detection sensitivity make UHPLC coupled with MS/MS an even more powerful analytical support in pharmacokinetic studies.



## ● GC-MS

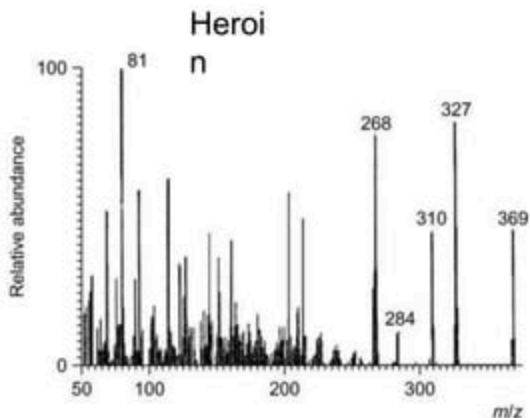
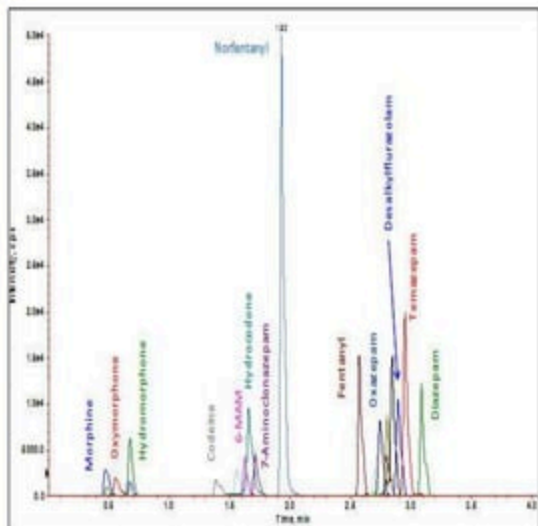
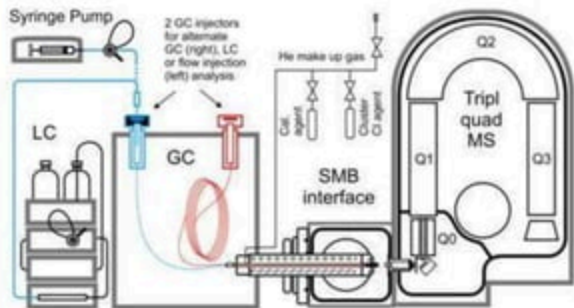
- The GC-MS analysis of polar compounds, such as metabolites, from biological matrices requires analytes extraction into a volatile organic solvent.
- time-consuming sample preparation including derivatization to become stable, volatile and amenable to the ionization technique.

## ● Capillary electrophoresis (CE)

- CE in many instances can have distinct advantages over HPLC in terms of simplicity, rapid method development, solvent saving and minimal sample requirement [10-30 nL injected] making this technique very interesting for rapid and practical analyses in the biomedical field.
- But have less sensitivity.

## ● Mass spectrometry

- Separate ion according to  $m/z$  ratio.
- Currently, the QQQ using single or multiple reaction monitoring is most often used for quantitative analysis of metabolites.
- SIM suffers from insufficient selectivity in comparison with MRM.
- And also much lower sensitivity.
- IT and TOF analyzers are also used for metabolite determination or use combined with QQQ (Qtrap, Q-TOF).



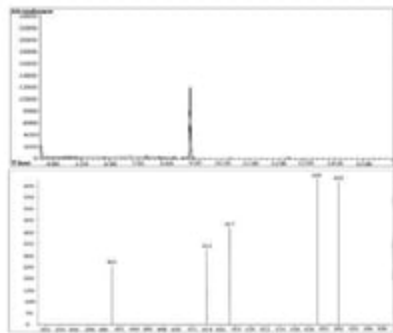
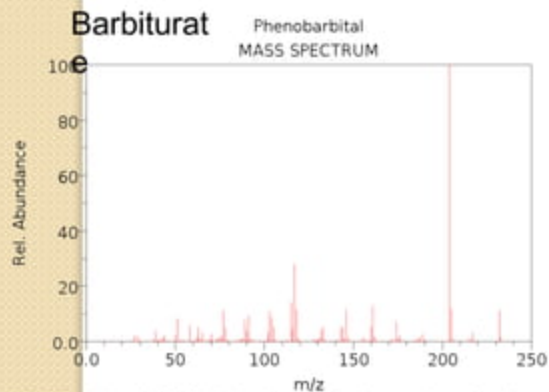
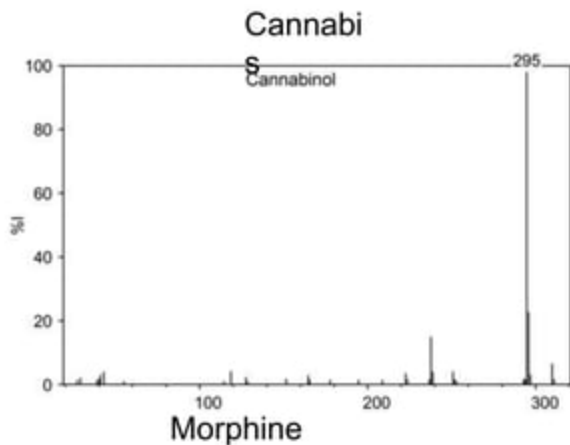
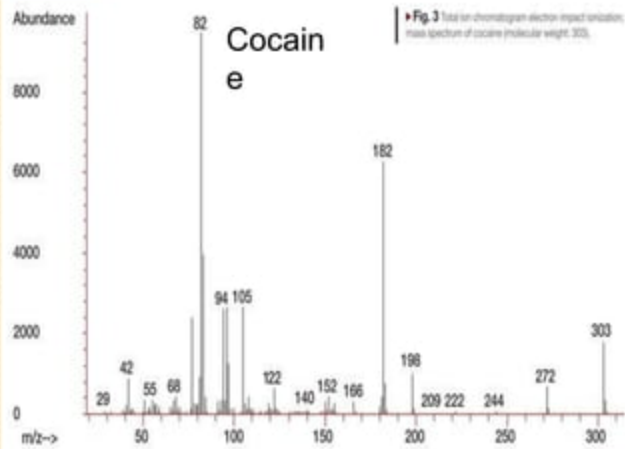


Figure 1. Determination of plasma morphine concentration through CG-MS. Chromatogram: morphine and morphine-D7 (internal standard 250ng/0.8 minutes, Run 10 minutes. Mass spectrum: mass 429 (morphine), mass 432 (morphine D7).

# Thin layer chromatography

## Acid mixture extract A



10 ml  
urine  
sampl  
e

1 ml of  
dil HCL → Shake for  
And 10 ml 5 min  
CHCl<sub>3</sub>



For 10 min



Then lower  
organic layer to  
tapered glass  
tube



Evaporate  
extract at  
60° C

## Basic mixture extract



10 ml  
urine  
sampl  
e

2 ml  
NH<sub>4</sub>Cl → Shake for  
and 5 min  
10ml  
CHCl<sub>3</sub>:  
propan-  
2-ol



For 10 min



Then lower  
organic layer to  
tapered glass  
tube

0.5 ml  
methanolic  
HCL



### 3. Purification of extracts of stomach contents

- Prior to the solvent evaporation stage, add 5 ml of aqueous sodium hydroxide solution to extract A, and 5 ml of aqueous hydrochloric acid to extract B.
- Shake on a mechanical shaker for 5 minutes, centrifuge in a bench centrifuge for 10 minutes and discard both organic layers.
- Add 5 ml of aqueous hydrochloric acid to the aqueous residue from extract A, and 5 ml of ammonium chloride buffer to the aqueous residue from extract B, and re-extract into chloroform or chloroform:propan-2-ol as in methods 1 and 2 above.

1	2	3	4	5	6	7	8
Acidic drugs mixture (10 µl)	Acidic sample extract (25 µl)	Basic drugs mixture (10 µl)	Basic sample extract (25 µl)	Basic drugs mixture (10 µl)	Basic sample extract (25 µl)	Pheno-thiazine mixture (10 µl)	Basic sample extract (25 µl)

10 cm

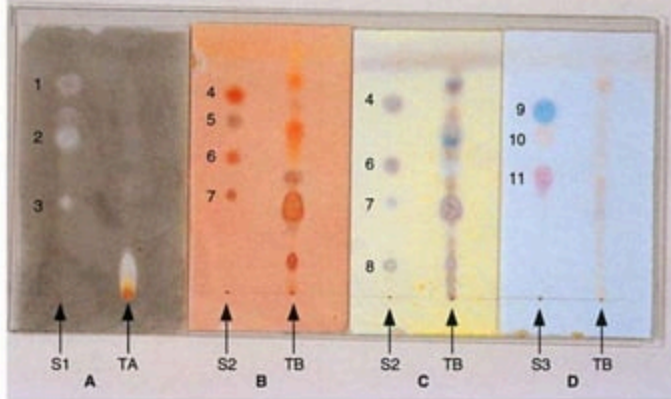
— Scored lines/Plate edges      - - - - Pencil line

#### Standards

All 1 g/l in chloroform:

- Acidic drugs mixture (amobarbital, mefenamic acid, phenobarbital, theophylline).
- Basic drugs mixture (amitriptyline, codeine, nicotine, nortriptyline).
- Phenothiazine mixture (perphenazine, trifluoperazine, thioridazine).

TLC visualization reagent-  
Mercurous nitrate  
Acidified indoplatinate  
FPN reagent  
Marquis reagent



Codeine and  
TLC visualization reagent methadone

- A Mercurous nitrate
- B Acidified  
indoplatinate
- C Mandelin reagent
- D Sulfuric acid

- 1 Amobarbital
- 2 Phenobarbital
- 3 Theophylline
- 4 Amitriptyline
- 5 Nicotine
- 6 Nortriptyline
- 7 Codeine
- 8 Mefanamic acid
- 9 Thioridazine
- 10 Trifluoperazine
- 11 perphenazine

# Analysis of stomach content

## Clinical interpretation

### Arsenic

Acute poisoning of Arsenic causes abdominal pain, vomiting, bloody diarrhoea, massive haemolysis.

### Mercury

Mercury vapors cause stomatitis, increased salivation, metallic taste, diarrhoea, pneumonitis, renal failure.

Mercury salts may cause gastric pain, vomiting.



# Reinsch Test

Qualitative test



Cu mesh  
clean in  
 $\text{HNO}_3$

Rinse the  
Cu with  
D.W.



Add 10 ml  
conc. HCl and  
20 ml test  
solution



For one  
hour



Fume  
cupboard

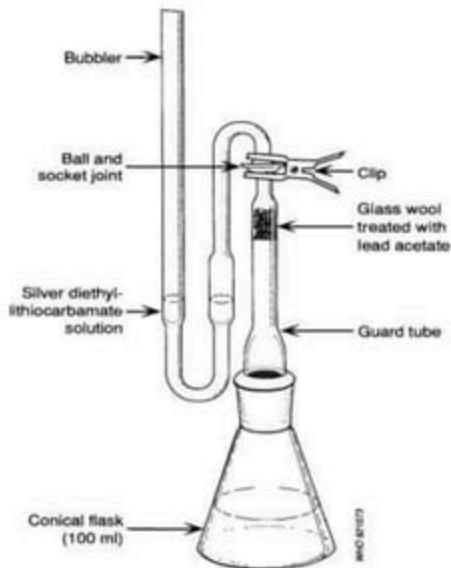
Cool and wash with  
D.W.



Results  
Dull black-  
Arsenic  
Silvery- Mercury

# Modified Gutzeit apparatus

## Quantitative test



Clean and dry with acetone  
Glass wool treated with Pb acetate  
Add 3 ml silver diethylthiocarbamate  
Into bubbler  
Add 2 g KI and 50 ml sample and swirl until dissolved. Then add 2 ml  $\text{SnCl}_2$  and 10 ml conc. HCl  
Add 10g granular zinc and quickly connect with bubbler. Kept for 45 min at RT. Absorbance of sol at 540 nm against blank and calculate arsenic concentration by previously prepared calibration graph.

# Cyanide

- Qualitative test

- Dissolve 1ml of sample in 2 ml NaOH. Add 2 ml ferrous sulphate and add sufficient HCl to dissolve ferrous hydroxide precipitate.
- Blue color indicates presence of cyanide.

## Quantitative test

Method

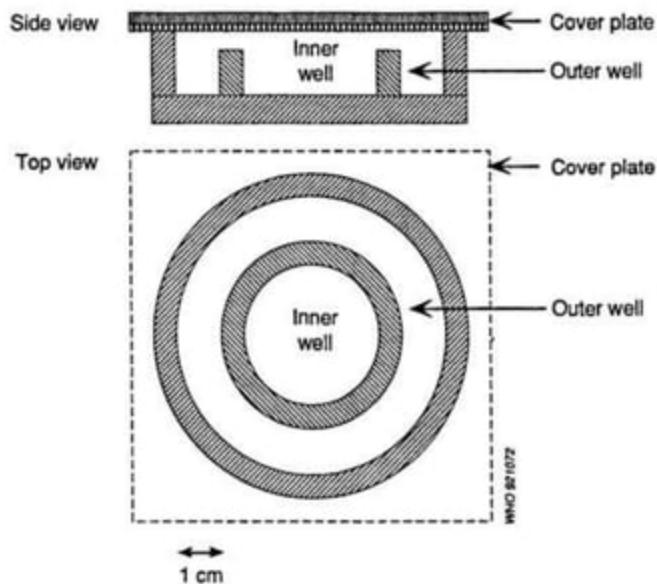
Take three microdiffusion cells and add to each of the centre wells

1 0.5 ml of p-nitrobenzaldehyde solution

2 0.5 ml of o-dinitrobenzene solution

3 0.5 ml of sodium hydroxide solution

Fig. 1. Conway microdiffusion apparatus



2. To the outer wells add 0.1 ml of:
  - purified water (cell 1);
  - potassium cyanide solution (cell 2);
  - test blood specimen (cell 3).
3. To each outer well add 0.5 ml of purified water and, on the opposite side of the outer well, 1.0 ml of dilute sulfuric acid.
4. Seal each well using silicone grease, and carefully mix the components of the outer wells.
5. Incubate at room temperature for 20 minutes and then add 1 ml of aqueous methanol (1:1) to the centre wells.
6. Transfer the contents of the centre wells to 5.0-ml volumetric flasks and make up to volume with aqueous methanol (1:1).

#### Results

The red coloration obtained with cyanide-containing solutions is stable for about 15 minutes. Measure the absorbance of the solutions from cells 2 and 3 at 560 nm against the purified water blank (cell 1).

Assess the cyanide ion concentration in the sample by comparison with the reading obtained from the standard.

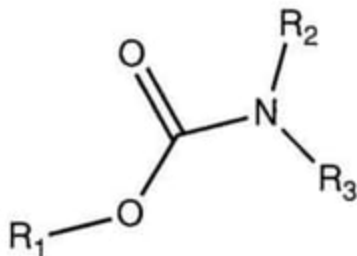
#### Sensitivity

Cyanide, 0.5 mg/l.

### Clinical interpretation

Causes ataxia, headache, anxiety, confusion, coma, metabolic acidosis, collapse, respiratory arrest.

# Carbamate pesticide



## Clinical interpretation

Exposure to carbamates may cause anorexia, abdominal pain, nausea, vomiting, diarrhoea, lacrimation, increased salivation, sweating, anxiety, ataxia and acute pulmonary oedema. Antidotal therapy with atropine may be indicated, but pralidoxime should not be used.

Acidify 1 ml sample by 0.5 ml dil. HCl & extract with 4 ml  $\text{CHCl}_3$

Shake for 5 min.



For 5 min.

Remove upper aq. Layer and filter  $\text{CHCl}_3$  extract by phase separating filter paper



Evaporate for 40° C

Dissolve residue in 1 ml methanol



Apply 0.1 ml furfuraldehyde



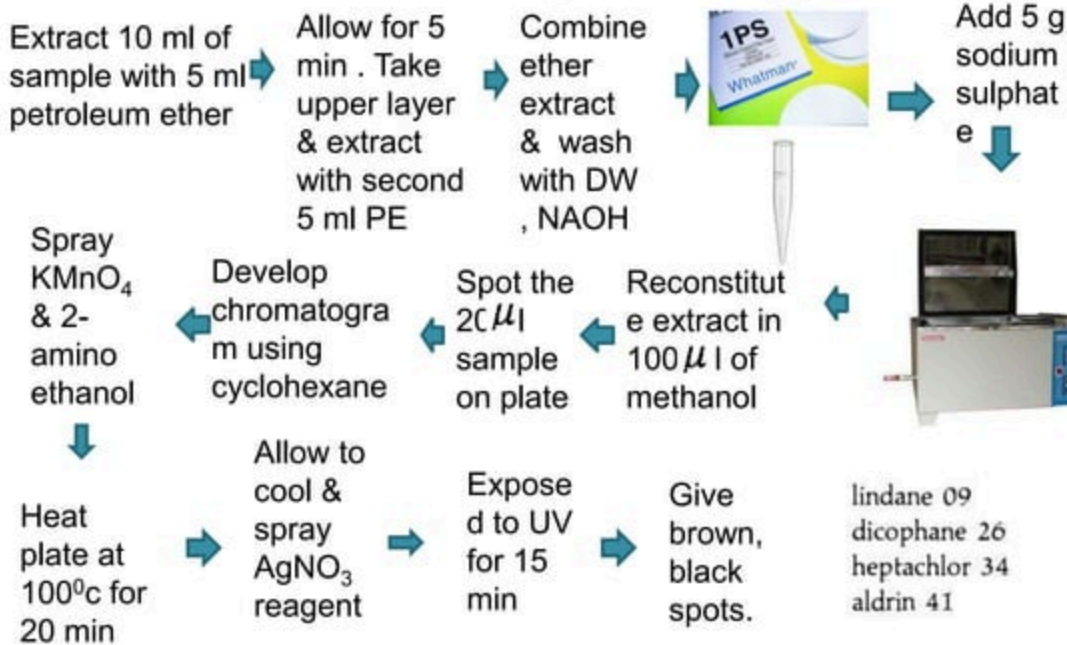
Expose paper for conc. HCl fumes for 5 min.

Carbamate gives black spot

# Organochlorine

## Clinical interpretation

Features of poisoning with organochlorine pesticides include vomiting, weakness and numbness of the extremities, apprehension, excitement, diarrhoea and muscular tremor, with convulsions and respiratory depression in severe cases. Treatment is symptomatic and supportive.





# Organophosphorous

Adjust the pH 7  
of 10 ml of  
sample with  
NaHCO<sub>3</sub>



Extract 10 ml of  
sample with 5  
ml MTB Ether



Allow for 5  
min . Take  
upper layer  
& extract  
with second  
5 ml MTB  
Ether



Follow  
same  
procedure  
as in OC



Spray the  
plate 4-(p-  
nitro  
benzyl)pyridin  
e



110°C for 30  
min



Allow to  
cool & spray  
Acetone:  
tetraethylen  
pentamine



Give  
purple  
spot

Clinical interpretation-  
May cause bronchorrhoea, Respiratory  
distress, nausea, muscle weakness ,  
paralysis

dimethoate	11
methidathion	40
malathion	42
dioxathion	47
propetamphos	49
bromophos	54
chlorpyrifos	58

**ANALYTICAL TOXICOLOGY REQUEST**

To: [Insert laboratory name, address and telephone no.]

Discuss special requirements BEFORE sending samples.

Doctor (PLEASE PRINT):

Telephone/bleep no:

Hospital address  
for report:

Signed:                      Date:

Patient:                      Other names:

Age/date of birth;                      Sex:

Consultant:                      Ward:

Reference no:

Sample type	Date	Time
Blood (10 ml heparinized)		
Urine (50 ml; catheter yes/no)		
Stomach contents (50 ml)		
Other (give details)		

Date/time of admission:

Date/time of ingestion or exposure:

Drugs prescribed or used in treatment:

Drugs/poisons claimed or suspected:

Clinical details/investigation required/priority:

# Reference

[www.who.int/ipcs/publications/training.../analytical\\_toxicology.pdf](http://www.who.int/ipcs/publications/training.../analytical_toxicology.pdf)

[jat.oxfordjournals.org/](http://jat.oxfordjournals.org/)

<http://dx.doi.org/10.5772/51676>

[en.wikipedia.org/wiki/Journal\\_of\\_Analytical\\_Toxicology](http://en.wikipedia.org/wiki/Journal_of_Analytical_Toxicology)



**THANK YOU**